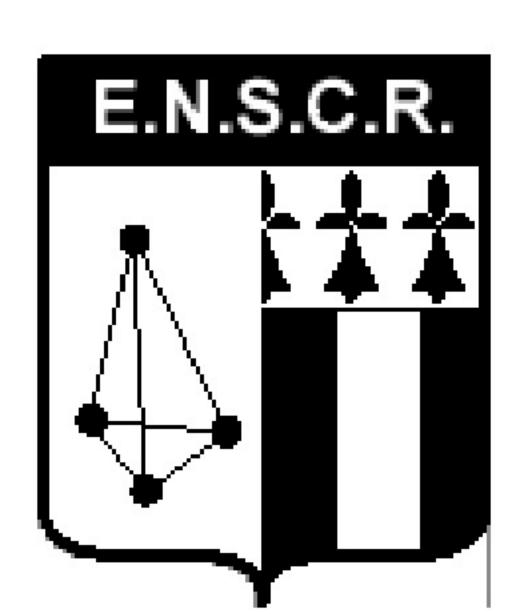
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Transition Metal Complexes for the Enantioselective Epoxidation of Olefines



A. Scheurer, a,c P. Mosset, c,* H. Maid, M. Spiegel, R. W. Saalfrank, A. R. Puchta, N. van Eikema Hommes, *

a Institut für Organische Chemie, Henkestr.42, D-91054 Erlangen

b Computer Chemie Centrum, Nägelsbachstr. 25, D-91052 Erlangen

c Lab. de Synthèse et Activations de Biomolécules, CNRS UMR 6052, ENSCR, Avenue du Général Leclerc, F- 35700 Rennes

Introduction

The asymmetric epoxidation of alkenes catalysed by chiral salen Mn(III) complexes **1-3** (*Jacobsen*, *Katsuki*, and *Bell*)^[1-3] is extensively studied in organic chemistry, since optically active epoxides are useful intermediates for the synthesis of many natural products.

Results

Starting from L-tartaric acid **4**, a series of new salen Mn(III)Cl complexes **6a-e** were prepared *via* diamines **5a-e**^[4,5] and studied as catalysts for the enantioselective epoxidation of unfunctionalised *Z*-alkenes (*ee* up to 69% were obtained). [6,7]

HO₂C CO₂H RO OR H₂N NH₂
$$t$$
-Bu t -

Compared to the literature catalysts **1-3**, stereochemically identical **6a-e** furnished epoxides with opposite configuration [(*S*,*S*)-**8**]. This unusual reversal of asymmetric induction is likely due to the co-ordination of the oxygen functionality in the diimine moiety to the metal centre. There is further evidence for an ether-metal interaction shown by ¹H NMR spectroscopy of **10** and the calculated model compound **11** (Figure 1).^[8,9]

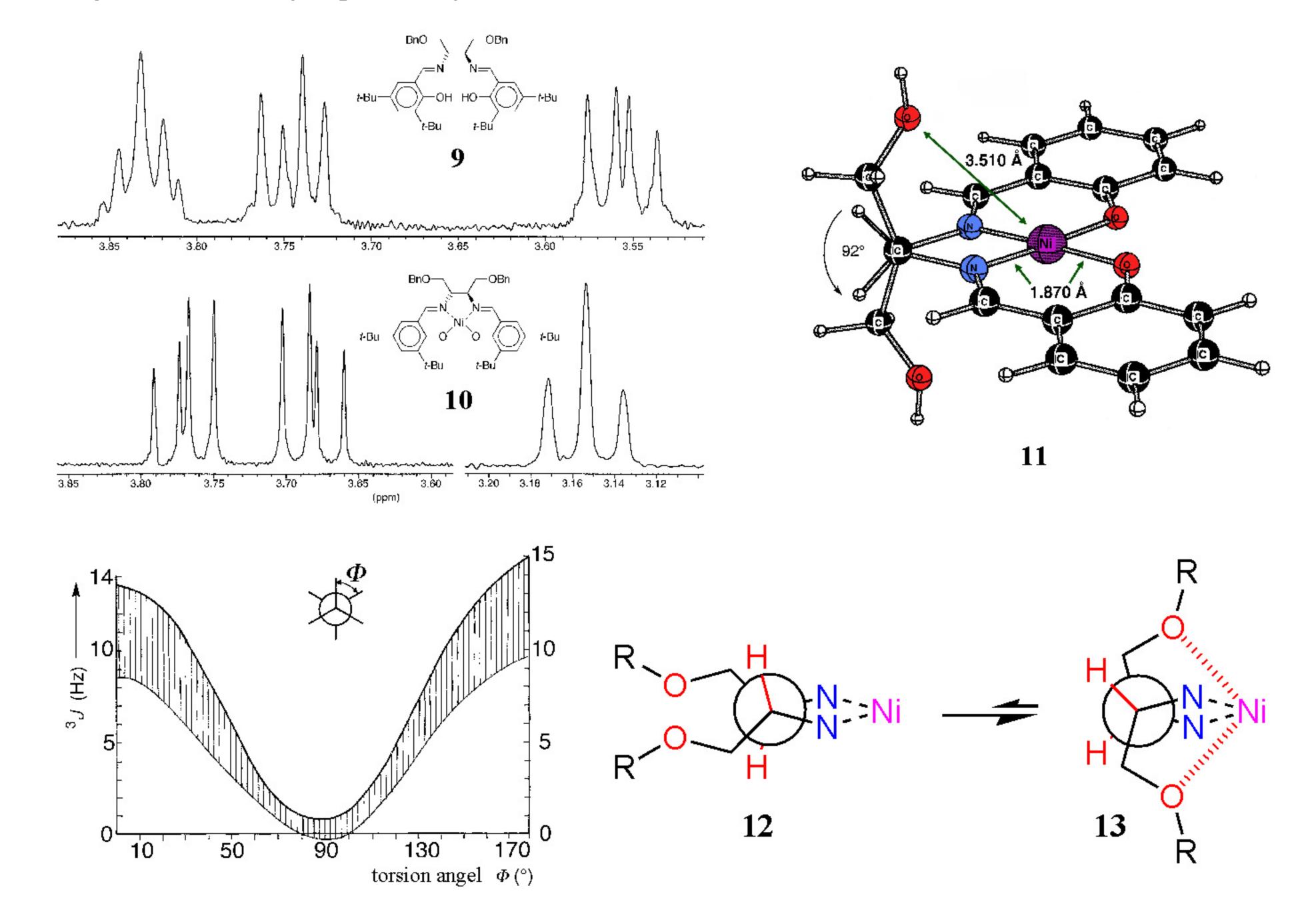


Fig. 1: ¹H NMR spectra of **9** and **10**, calculated structure **11**, *Karplus*-curve, and the possible Newman projections **12**,**13** for salen Ni(II) complex **10**.

For the salen Ni(II) complex **10**, the ¹H NMR spectrum showed an AA'BB'CC'-spin system with a vicinal coupling constant ${}^3J_{\rm CC'}\approx 0$ Hz (ligand **9**: ${}^3J_{\rm CC'}=4.3$ Hz). Therefore, following the *Karplus*-curve, a dihedral angle Φ of about 90° between the vicinal hydrogens in the backbone of **10** is expected. This is in good accordance with the proposed Newman projection **13** and the calculated structure of model compound **11** (Φ = 92°).

Compared to 1 and 3, the influence of a larger carbocyclic substituent in the chiral unit was studied. Starting from trans-1,2-cyclooctanediol 14, the new salen Mn(III)Cl complexe 16 was generated via diamine 15 (resolution with L-tartaric acid afforded 15 with ee 87%).^[10] The enantioselective epoxidation catalysed by 16 of chromenederivatives 7 led to epoxides 8 with low optical yield ($ee \le 25\%$).

In Figure 2 the calculated model compound^[9,11] and the crystal structure of **16** are represented. The eight-membered ring in the backbone of **16** is twisted towards the molecular plane which led probably to the poorer enantioselection.

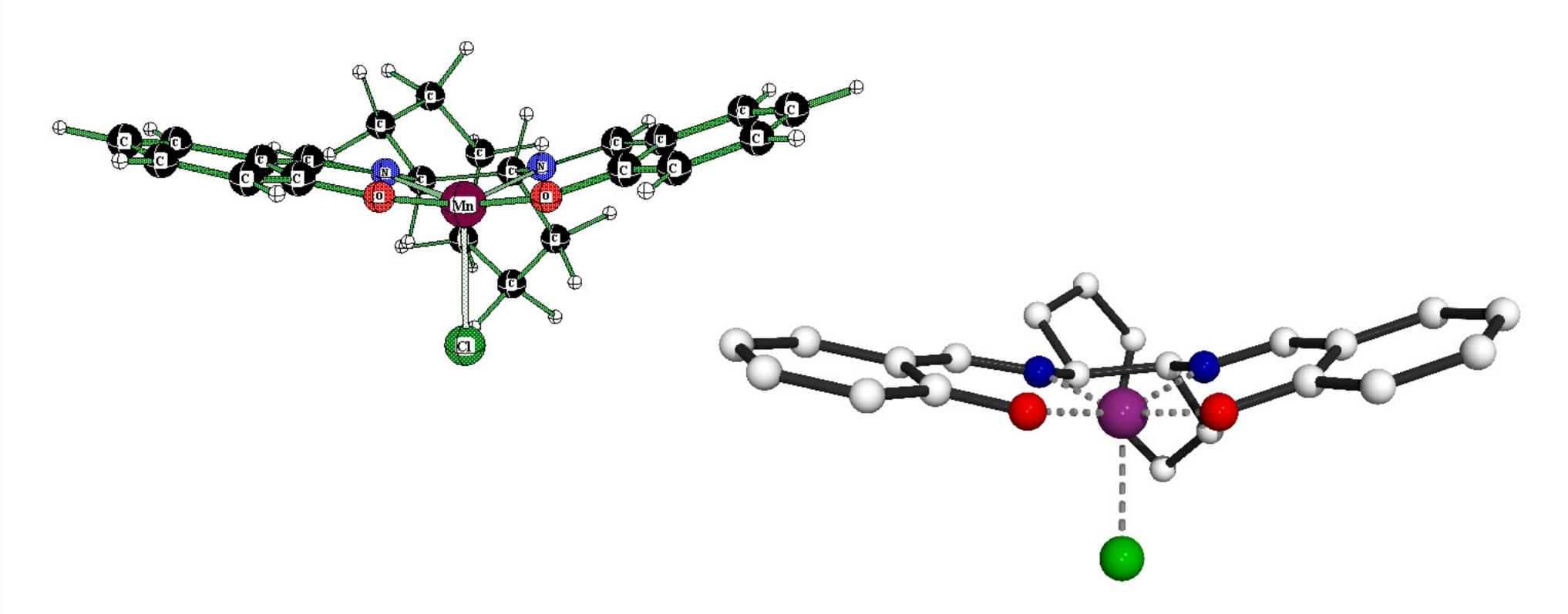


Fig. 2: Calculated structure of model compound (left) and crystal structure of 16 (right).

Conclusion

The oxygen functionality of the diimine moiety in complexes **6a-e** coordinates to the metal centre. This leads to a *pseudo-enantiomeric* conformation of **6a-e** which causes the inversed enantioselection. The introduction of a eight-membered ring in **16** does not support the enantioselectivity.

Literature

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